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It is easier to remove an electron from Na^+ than from Na .; The first ionization energy of fluorine is greater than the first ionization energy of oxygen. ? All particles with the electron configuration

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[Ar]4s 2 have the same ionization energy.; As the atomic number increases within a group of the representative elements, the tendency is for first ionization energy to increase.

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Ionic and ... - Study.com

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7 1. When ignited, a uranium
compound burns with a green
flame.

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Fall 2014 Instructor: Michael C. Biewer biewerm@utdallas.edu, BE 3.326 Syllabus. Lecture Notes: The following material can be downloaded as pdf files. Material for test #1 . Chapter 1 Chapter 2 Chapter 3 Chapter 5 . Material for test #2 . Chapter 4

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operators often broadcast on the 6-meter band. The frequency of this electromagnetic radiation is

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between a positive nucleus and negative electrons, or positive and negative ions.

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