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*transfers of energy
between chemical
and physical systems.
We learned that the
first law of
thermodynamics is
the ...*

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questions 1 and 2,
illustrate your
answers with simple
drawings of the
particles that form the
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Chemical Reactions.
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Reactions energy ...*

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*7 Chemical Formulas
and Chemical
Compounds
AP Chemistry Review
Questions - Atomic
Structure and
Periodicity ... 7;
Excluded How many
electrons in an atom
can have $n = 5, l = 3$?*
*? 2 ? 6 ? 10 ? 14 ?
None of these;*

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Researchers at Lawrence Berkeley National Lab have recently formed a new synthetic element with atomic number 118 and mass number 293. Which of the following elements would have ...

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Chemistry 101

ANSWER KEY

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REVIEW

QUESTIONS Chapter

**4 1. Identify each of
the following**

**substances as a non-
electrolyte (NE), weak
electrolyte (WE), or
strong electrolyte
(SE): a)**

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CHAPTER 7 REVIEW

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*Chemical Formulas
and Chemical
Compounds*

SECTION 1 SHORT

ANSWER Answer the following questions in the space provided. 1.

c In a Stock system name such as iron(III) sulfate, the Roman numeral tells us (a) how many atoms of Fe are in one formula unit. (b) how many

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sulfate ions can be attached to the iron atom. (c) the charge on ...

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*Chemistry Chapter 7
Lesson 7.1*

Precipitation

*Reactions - a process
in which soluble ionic
reactants yield an*

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insoluble product that falls out of the solution, thus removing some of the dissociated ions in the solution

Acid Base Reactions

0 processes in which an acid reacts with a base to yield water plus an ionic compound called a salt

Redox Reactions - a process in which

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one or more electrons

...

*Ch 7 Ionic and
Metallic Bonding
Section Review 7.1
Part A Completion .
valence electrons 2.
group electron dot
structures 3. octet rule
4. 5. cations 6. anions
Halide ions 8. 9. gain
charges 10. 15. AT
16. NT 22. a 23. c*

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True/False 13. ST 14.

NT Matching 20. g 21.

f Part B 11. NT 12. AT

Part C 18. 19. b d e .

Section 7.3

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ANSWER KEY 1

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QUESTIONS Chapter

4 1 ...

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