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Worksheet 1.3:

Reactions

Complete the

following

reactions,

identify the

reaction type

and balance the

equation. (3

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marks each; 15
marks total) 1)

mercury (II)
oxide is broken
down into its
elements by
heating. 2) a
nickel strip is
placed in a gold
(III) sulfate
solution 3)
phosphoric acid
reacts with iron
(III) oxide.

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two solutions
are mixed a
chemical
reaction can
occur between
the dissolved
solutes and we
need to be able
to
quantitatively
describe these
reactions.

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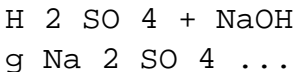
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Given the following reaction: (hint: balance the equation first)



If 36.7 mL of HCl solution is needed to react with 43.2 mL of a 0.236 M NaOH, what is the concentration of

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are Chem 10

level problems.

They are

provided to

assist your

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in Chp 4 of the

Zumdahl

textbook. Note

that Chem 11

problems will be

more involved

and more

rigorous than

these!

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solving these
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problems is to
set up the
problem so that
the units
cancel. When the
volume of a
solution is
multiplied by

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the molarity of a solution the resulting units are moles. A balanced equation allows us to convert from moles of a known substance to moles of an unknown.

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Titrations, etc.

Example

Problems!page 2

Titration

Shortcut (Only

for reactions

with a 1:1

ratio) $V_1 M_1 =$

$V_2 M_2!$ (units

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match) Calcium ions or

Magnesium ions

react with a

titrant called

EDTA in a 1: 1

ratio. 1. 10.3

mL of 9.75 mM

EDTA solution is

used to titrate

15.0 mL of a

calcium

solution. What

is the

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day. As a

community, we

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together and

support each

other. Iron

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the following

solutions

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problems: 1. How

many grams of

silver chromate

will precipitate

when 150. mL of

0.500 M silver

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nitrate are
added . to 100.
mL of 0. ...
titration of no
fewer than 15.0
mL of 0.100 M
NaOH? Both of
the hydrogen's
from oxalic acid
are

Mr .

Christopherson /

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a solution is a
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